

# Comparative study on perovskite solar cells using P\_ZnO, Al\_ZnO and In\_ZnO as ETMs by SCAPS-1D

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## Abstract

This study uses the SCAPS 1D software to analyze solar cells with lead iodide perovskite (CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub>) as the active material and three different types of ZnO doping: undoped (P ZnO), aluminum-doped (Al ZnO), and indium-doped (In ZnO) as the electron transport layer (ETL). This study aims to investigate the effects of charge carrier density on the J-V characteristics and electrical properties (Jsc, Voc, FF, Eff) of a solar cell structure made up of FTO/ETL/CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub>/CuInSe<sub>2</sub>/Au. Gold makes up the back contact, and tin oxide doped with fluorine (FTO) makes up the front contact. These two compounds have work roles are 5.47 eV and 4.4 eV respectively. Zinc oxide, both undoped and doped, makes up the electron transport layer, whereas methyl ammonium lead iodide makes up the absorber layer. Thin sheets, indium diselenide, and copper make up the hole transport layer. An optimal perovskite layer was obtained by decreasing the electron transportation layer's thickness (ETL) from 500 to 100 nm. For the In ZnO ETL, this optimization was accomplished at 300K working temperature. Reduced ETL thickness was shown to result in higher efficiency (above 27%), as well as higher fill factor (above 87%). With a Voc value of 1.180V, a FF value of 87.55%, and a J<sub>sc</sub> value of 26.276mA/cm<sup>2</sup>, the greatest efficiency performance of 27.15% was found. Using an absorber layer made of CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub> with a thickness of 800 nm and indium-doped oxide of zinc (In ZnO) is the layer that transports electrons, this performance was achieved. The results are obtained at a constant irradiance level of 1000 W/m<sup>2</sup>, under the AM1.5G spectrum.

**Keywords:** CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub>. ZnO. SCAPS-1D. Carrier concentration. Thickness. Aluminum doped ZnO. Indium doped ZnO.

## 1. Introduction

Owing to its potential for applications requiring high-performance solar cells at a reasonable cost, perovskite organic-inorganic material has garnered significant interest when employed as an active element in solar cells [1]. This material is a good contender for next developments in solar cell technology because of its encouraging properties [2]. Copper indium diselenide is widely used in the fabrication of thin-film solar cells (CuInSe<sub>2</sub>) and oxide of zinc (ZnO) as raw materials. In particular, ZnO is the layer responsible for transporting electrons, while CuInSe<sub>2</sub> is responsible for transporting holes in perovskite solar cells that use methylammonium lead iodide (CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub>) as the active component [3,4].

The band gap that defines the material is about 1 eV [5] and 3.26 eV [6] and displays a direct energy transition because of its strong absorption coefficient  $(10E+5 \text{ cm}^{-1})$  [7,8]. Furthermore, the cheap cost of its production processes—such as electrodeposition, sol-gel, and spray pyrolysis—is a distinguishing feature. ZnO and CuInSe<sub>2</sub> are important materials when considering renewable energy and their applications [9]. The compound CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub> has a strong absorption coefficient ( $\alpha$ ) [10,11] and the gap in the band of the straight transition about of 1.66 eV [10]. It is therefore utilized as the active layer in p-i-n and n-i-p designs. The efficiency of solar cells that use methyl ammonium lead iodide has significantly increased recently [12,13], considering level of illumination of 1000 W/m<sup>2</sup>.

Various metals can be doped into zinc oxide, including Indium, Gallium and Aluminium, to enhance its electrical, optical, and structural characteristics [14-17]. Oxide of Zinc (ZnO) thin transparent conductive films, both doped and undoped, are important for electrical and optical devices, particularly in the context of solar cells [15]. Here, we report the findings of a computerized model simulation of a solar cell that uses methylammonium. (MA) lead iodide as the active material. The simulations were conducted using a one-dimensional SCAPS-1D simulator. The photovoltaic cell structure's electron transportation layer (ETL), which is made up of FTO/ETL/CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub>/CuInSe<sub>2</sub>/Au, was composed of pure ZnO (P ZnO), Indium doped ZnO (In ZnO), and Aluminum doped ZnO (Al ZnO). We investigated the effects of carrier concentration and thickness (W) depending on the solar cell's efficiency. The acquired data show a fill factor value greater than 87%.

# 2. Details of simulation

In this title, we employ CuInSe<sub>2</sub> as the hole transport layer (p-type), CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub> as the absorber layer (p-type), and undoped ZnO, aluminum-doped ZnO, and indium-doped ZnO as the electron transport layers (n-type). Furthermore, we utilize gold (Au) as a back contact for an analytical function of ( $\phi$ ) of 5.47 eV [18] and SnO<sub>2</sub>: F (FTO) as the front contact with analytical function ( $\phi$ ) of 4.4 eV [19], so it is illustrated in Figure 1.



Figure1: representation of the structure of CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub> solar cell.

The Electronics and Information Systems Department (ELIS) at Gent University in Belgium is where the SCAPS program originated [1]. As mentioned earlier, the calculations are executed through the resolution of a trio of essential equations, specifically the formula for Poisson (equation 1) and the continuation formulas that regulate the conduct of holes and electrons. (equations 2 and 3) [20, 21].

$$\frac{d}{dx}\left(\varepsilon(x)\frac{d\psi}{dx}\right) = q[p(x) - n(x) + N_D^+(x) - N_A^-(x) + p_t(x) - n_t(x)]$$

$$\frac{1}{j}\frac{\partial J_p}{\partial x} + R_p(x) - G(x) = 0$$
(1)
(2)

$$\frac{1}{i}\frac{\partial J_n}{\partial x} + R_n(x) - G(x) = 0 \tag{3}$$

Where:

- ε permittivity;
- $\psi$  potential electrostatically.
- q charge of electron.
- n, p concentration of electrons and holes.
- n<sub>t</sub>, p<sub>t</sub> confined electron (negative charge), hole (positive charge).
- $N_D^+$ ,  $N_A^+$  ionized doping concentration of donor and acceptor.
- $R_n(x)$ ,  $R_p(x)$  electrons and holes recombination rates.
- G(x) generation rating.
- J<sub>n</sub> and J<sub>p</sub> electron and hole current densities.

All parameters of simulation are selected in table 1 from experimental and other theoretical results [2][4][6][9][22-26], the absorption coefficient ( $\alpha$ ) of CH3NH3PbI3 prepared by [2], while the absorption coefficient of ZnO and CuInSe2 by SCAPS 1d.

	P_ZnO	Al_ZnO	In_ZnO	CH3NH3PbI3	CuInSe <sub>2</sub>
W (nm)	100-500	100-500	100-500	800	200
Eg (eV)	3.26	3.269	3.28	1.54	1.07
ξr	9	9	9	6.5	9
Xe (eV)	4.1	4.1	4.1	3.93	4
μn (cm²/Vs)	100	100	100	2	$5.000 \times 10^{-4}$
$\mu_p$ (cm <sup>2</sup> /Vs)	25	25	25	2	$5.000 \times 10^{-4}$
$N_{c}$ (1/cm <sup>3</sup> )	$2.22 \times 10^{18}$	2.22	2.22	$2.2 \times 10^{18}$	$2.2 \times 10^{18}$
		$\times  10^{18}$	$\times  10^{18}$		
$N_v(1/cm^3)$	$1.8  imes 10^{19}$	$1.8  imes 10^{19}$	$1.8  imes 10^{19}$	$1.8  imes 10^{19}$	$1.8  imes 10^{19}$
Nd (1/cm <sup>3</sup> )	$2 \times 10^{13}$	7.589	$1.2 \times 10^{20}$	0	0
		$\times  10^{18}$			
N <sub>a</sub> (1/cm <sup>3</sup> )	0	0	0	$2 \times 10^{17}$	$2 \times 10^{16}$
Vth <sub>e</sub> (cm/s)	$1 \times 10^{7}$	$1 \times 10^{7}$	$1 \times 10^{7}$	$1 \times 10^{6}$	$1 \times 10^{7}$
Vth h (cm/s)	$1 \times 10^{7}$	$1 \times 10^{7}$	$1 \times 10^{7}$	$1 \times 10^{6}$	$1 \times 10^{7}$

Table The simulation parameters of the input me on the Chiji (11) big solar ce	Table1– The simulatio	parameters of	f the input file o	n the C	H3NH3PbI3	solar cell.
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#### 3. Results and discussions

# 3.1 Effect of carrier concentration of ETL on CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub> cells

In the following subsection, we examine the zinc oxide's  $N_d$  doped percentage concentration, we use undoped ZnO (P\_ZnO), Aluminum doped ZnO (Al\_ZnO) and Indium doped ZnO (In\_ZnO) in this structure "FTO/ (P\_ZnO, Al\_ZnO, In\_ZnO)/CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub>/CuInSe<sub>2</sub>/Au".

Undoped ZnO prepared by S.Benzitouni et al in 2018 [6], the obtained result 3.26eV as band gap energy (Eg) and carrier concentration (N<sub>d</sub>) equal to  $2*10^{13}$ cm<sup>-3</sup>. The Al doped ZnO prepared by A.Nakrela and co-authors [22], he obtained 3.269eV as band gap energy and the N<sub>d</sub> equal to  $7.589 \times 10^{18}$  cm<sup>-3</sup>. The In doped ZnO prepared in 2014 by [23], he obtained 3.28eV as Eg and the N<sub>d</sub> equal to  $1.2 \times 10^{20}$  cm<sup>-3</sup>(Table 2).

Structure	Nd	Eg	W	Eff	FF	Jsc	Voc
	(cm <sup>-3</sup> )	(eV)	(nm)	(%)	(%)	(mA/cm <sup>2</sup> )	<b>(V)</b>
P_ZnO/CH3NH3PbI3/	$2 \times 10^{13}$	3.26	488	25.43	83.81	25.723	1.179
CuInSe <sub>2</sub>							
Al_ZnO/	$7.589 \times 10^{18}$	3.269	100	27.02	87.25	26.244	1.180
CH3NH3PbI3/CuInSe2							
In_ZnO/	$1.2 \times 10^{20}$	3.28	197	26.90	87.55	26.037	1.179
CH <sub>3</sub> NH <sub>3</sub> PbI <sub>3</sub> /CuInSe <sub>2</sub>							

# Table2 - cell performances on perovskite solar cell.

The impact of the concentration of transporters N<sub>d</sub> of P\_ZnO, Al\_ZnO, and In\_ZnO with thickness (W) and band gap (Eg) variables was prepared by experimental results (see table 2). N<sub>d</sub> was varied from  $2 \times 10^{13}$  to  $1.2 \times 10^{20}$  cm<sup>-3</sup>. The efficiency increased from 25.43 to 27.02% as well as the same the fill factor (FF) coming from 83.81 to 87.55%.

Table 2 presents the data illustrating the fluctuations in carrier concentration (N<sub>d</sub>) under shortcircuit conditions. The current density at short circuit (J<sub>SC</sub>) and the open circuit voltage (V<sub>oc</sub>) were investigated in this study. It was shown that J<sub>sc</sub> rose from 25.723mA/cm<sup>2</sup> for P\_ZnO to 26.244mA/cm<sup>2</sup> for Al-doped ZnO, while it reduced to 26.037mA/cm<sup>2</sup> for indium-doped ZnO. However, the change in V<sub>oc</sub> was found to be minor. The decrease in resistivity with increased ETL doping concentration can be attributed to this phenomenon [27].

In an investigation led by Adhikari and al. in 2016, the researchers investigated the impact of the dopant concentration of electron transport materials (ETMs) about the efficiency of solar cells. The results of their study indicated that the spiroMeOTAD/MAPbI<sub>3</sub>/ZnO structure achieved an efficiency of 22.49% [28]. The incorporation of aluminum (Al) and indium (In) into zinc oxide (ZnO) results in highly favorable perovskite solar cell characteristics, specifically yielding a maximum fill factor value above 87%.

#### 3.2 Role of thickness for each case on cell performance

The thickness (W) of doped and undoped ZnO varied from 100 to 500nm, while keeping the absorber (CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub>) and HTL (CuInSe<sub>2</sub>) layer thickness at 800 and 200nm respectively.

Figure 2 Clarifies the connection between CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub> solar cells' performance and the ETL layer thickness. It is observed that the efficiency experiences a minor decline from 26.08% to 25.41%, 27.02% to 26.43%, and 27.15% to 26.54% for P\_ZnO, Al\_ZnO, and In\_ZnO, respectively. The fill factor experiences a reduction from 84.53% to 83.77% in the case of P\_ZnO. However, in the case of Al\_ZnO, it was observed that the fill factor (FF) exhibited a little rise from 87.25% to 87.28% with the electron transport layer (ETL) the thickness of was increased (Figure 3). The effects involve altering the electron transport layer's thickness (ETL), namely In\_ZnO and Al\_ZnO, on the fill factor (FF) is negligible. The J<sub>sc</sub> values for P\_ZnO, Al\_ZnO, and In\_ZnO structures (Figure 4) fall from 26.144 to 25.719 mA/cm<sup>2</sup>, 26.244 to 25.664mA/cm<sup>2</sup>, and 26.276 to 25.694mA/cm<sup>2</sup>, respectively. Conversely, the V<sub>oc</sub> values (Figure 5) exhibit negligible changes. Each outcome aligns with the study carried out by Jayakumar et al [29] and Aseena et al [30].



Figure 2: influence of thickness for each case on efficiency.



Figure 3: influence of thickness for each case on fill factor.



Figure 4: influence of thickness for each case on short circuit current density (Jsc).



Figure 5: Influence of thickness for each case on open circuit voltage.

The optimized layer thickness of ETL is 100nm which corresponds to a maximum efficiency of 26.08%, 27.02% and 27.15% for P\_ZnO, Al\_ZnO and In\_ZnO respectively.

# 4. J-V Characteristic and quantum efficiency

Table 1 shows the preset doping density, absorber thickness, and hole transport layer (HTL) thickness. However, the electron transport layer's (ETL) thickness is fixed at 100 nm in each situation. Each of the three structures J-V properties, namely P\_ZnO, Al\_ZnO, and In\_ZnO/CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub>/CuInSe<sub>2</sub>, exhibit a high degree of similarity, as seen in Figure 6. Figure 7 demonstrates the quantum efficiency, indicating that opting for a broader energy gap for the electron transport layer (ETL) results in a notable enhancement in photon absorption within the ultraviolet (UV) spectrum [31].



Figure 6: (P\_ZnO, Al\_ZnO, In\_ZnO) as ETLs J-V characteristics



Figure 7 : CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub> solar cell quantum efficiency (QE)

#### 5. Conclusions

The main object of this study is to find out what the transport layer of electron (ETL) does for the structure of a  $CH_3NH_3PbI_3$  solar cell that uses P\_ZnO, Al\_ZnO, and In\_ZnO as electron transport materials (ETMs). We explore the influence of carrier concentration N<sub>d</sub> and thickness (w) of ETMs on cell performances. Al and In-doped ZnO offer good perovskite solar cell characteristics, but there is a little variation in findings between doping with Al and In. A statistical examination was conducted to compare the efficiency of devices and the level of quality of the perovskite layer. When we assess, there is a significant enhancement observed in the steady-state Fill factor to 87.55% and efficiency equal to 27.15%.

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